

EGN 3365: Materials in Engineering

- Introduction

Course Objective...

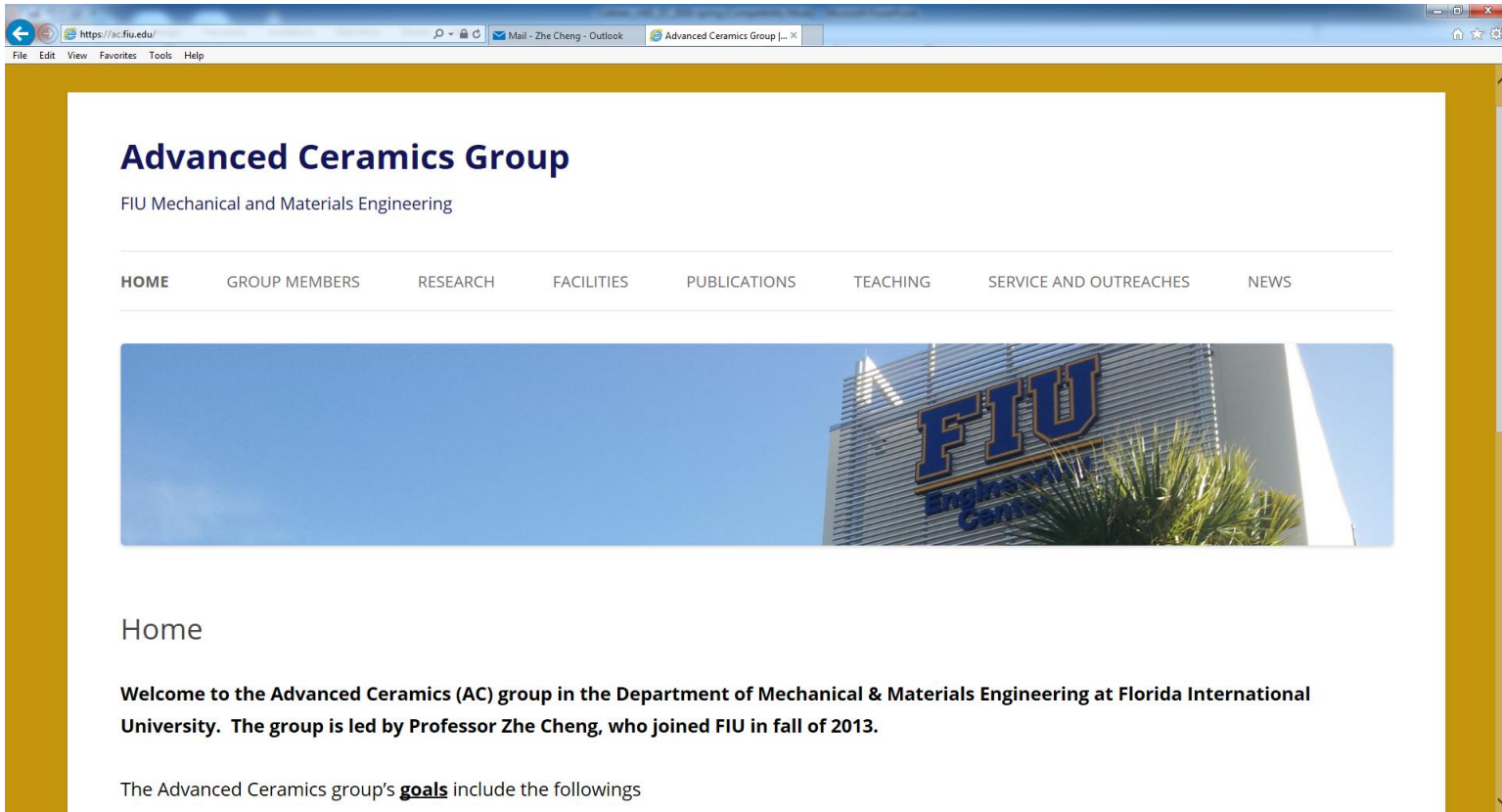
Introduce fundamental concepts and methods in materials engineering (and science)

- **Instructor: Dr. Zhe Cheng**
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 - Email: zhcheng@fiu.edu
 - Office: EC3172
- **Course website (for lecture slides and review information):**
<https://ac.fiu.edu/teaching/egn3365/>
- **Instagram chat group:**
- **Prerequisites:**
 - CHEM1045 General Chemistry I;
 - MAC 2311 Calculus I;
 - PHY 2048 Physics with Calculus I



More About Dr. Zhe Cheng

- **Education & Experiences:**
 - PhD in Materials Science & Engineering, Georgia Tech 2008
 - Research scientist at DuPont 2008-2013
- **Research group website:** <https://ac.fiu.edu>



The image shows a screenshot of a web browser displaying the homepage of the Advanced Ceramics Group at Florida International University. The browser's address bar shows the URL <https://ac.fiu.edu/>. The website has a yellow header with the title "Advanced Ceramics Group" and the subtitle "FIU Mechanical and Materials Engineering". Below the header is a navigation menu with links for HOME, GROUP MEMBERS, RESEARCH, FACILITIES, PUBLICATIONS, TEACHING, SERVICE AND OUTREACHES, and NEWS. A large banner image features the FIU logo and the text "FIU Engineering Center". Below the banner, the text reads "Home" followed by a welcome message: "Welcome to the Advanced Ceramics (AC) group in the Department of Mechanical & Materials Engineering at Florida International University. The group is led by Professor Zhe Cheng, who joined FIU in fall of 2013." The page also mentions the group's goals.



Course Feature

- **Interdisciplinary & broad**
- **Raising awareness & common sense**
- **Not exactly a specific tool/skill based course**
- **Every material/application is unique**

This course will help you to know:

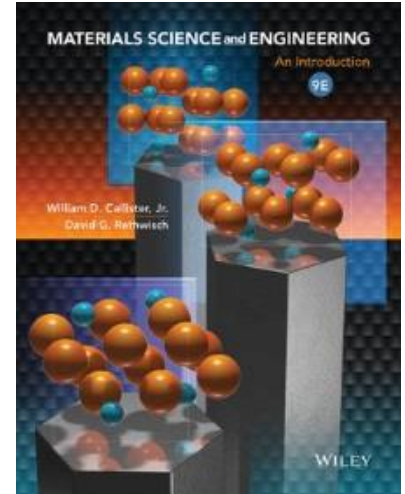
- basic concepts in materials engineering
- major categories of materials
- considerations in materials selection/design/processing



Textbook & Other Materials

- **Materials Science and Engineering
An Introduction,
William D. Callister, Jr. and David G Rethwisch,
John Wiley & Sons Inc, 9th ed. (2013)**

[Any other earlier versions are OK!](#)

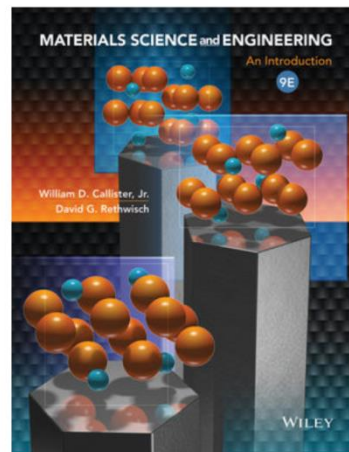
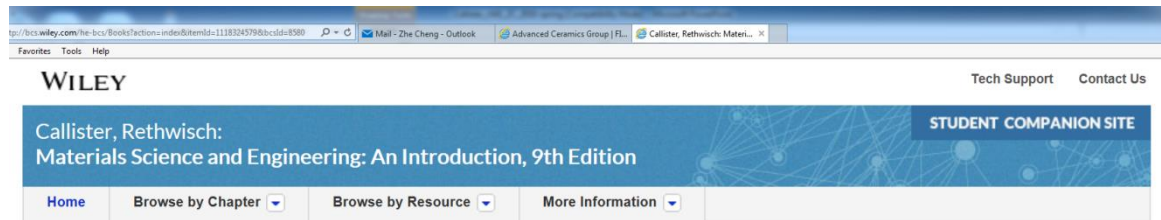


- **Wiley links for book purchase/rental
<http://www.wiley.com/WileyCDA/WileyTitle/productCd-EHEP002505.html>**
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Wiley Student Companion Site

- <http://bcs.wiley.com/he-bcs/Books?action=index&itemId=1118324579&bcsId=8580>
- Extra contents:
 - Concept check answers
 - Virtual Materials Science & Engineering (VMSE)
 - Users can manipulate molecules and crystals to better visualize atomic structures of crystals
 - Polymer repeat units and molecules
 - Diffusion computations
 - Student Lecture Slides
 - Case Studies



Welcome to the Student Companion Site for
Materials Science and
Engineering: An Introduction,
9th Edition

Welcome to the Web site for *Materials Science and Engineering: An Introduction, 9th Edition* by William D. Callister. This Web site gives you access to the rich tools and resources available for this text. You can access these resources in two ways:

1. Using the menu at the top, select a chapter. A list of resources available for that particular chapter will be provided.
2. Using the menu at the top, select a resource. This will allow you to access a particular resource section. You will then have the option of selecting resources within the section or going directly to a specific chapter.



Grading

- Homework (10 points)
- Attendance (8 points, in the form of **class exercises**)
- Group presentation (7 points)
- Three **closed-book** exams (25 points each for each)
- Final Grades:
 - A: ≥ 90 ; A-: 87-89.9;
 - B+: 84-86.9; B: 80-83.9; B-: 77-79.9;
 - C+: 74-76.9; C: 67-73.9;
 - D: 60-66.9;
 - F: < 60

Past statistics

- 2017 spring: **Median: 78.3**; Standard deviation: 18.0; Highest: 96.8
- 2016 fall: **Median: 77.3**; Standard deviation: 17.1; Highest: 98.1
- 2016 spring: **Median: 74.8**; Standard deviation: 12.2; Highest: 97.0
- 2015 fall: **Median: 80.1**; Standard deviation: 19.1; Highest: 94.5
- 2015 summer: **Median: 81.8**; Standard deviation: 11.6; Highest: **95.1**
- 2015 spring: **Median: 73.6**; Standard deviation: 21.3; Highest: 91.7
- 2014 Fall: **Median: 76.7**; Standard deviation: 9.1; Highest: 90.8



Course Policy

- Attendance required, checked in the form of class exercise
- Cell phones and other device on silent mode
- Can discuss homework problems, but must independently finish them
- Exams will be closed-book, 1 letter page (2-sided OK) formula/concept sheet allowed
- Grade discrepancies – resolve within the next business day
- **“Make-up” exams only for real, life-threatening emergency with valid proof**
- Accommodate disability/religious holidays
- **NO cheating or plagiarizing in ANY form**



Guidelines on Group Presentation

- A 12 min max presentation (~12 slides) about materials evolution/development of interest to you
 - It should be based on one of the two formats below:
 - The evolution/development of a **SPECIFIC MATERIAL** (e.g., man-made diamond), or
 - The evolution of materials used for a **SPECIFIC APPLICATION** (e.g., materials for solar cell light absorber)
 - It should have most of the following (but NOT necessary all)
 - Introduction
 - Materials development (may have multiple generations)
 - » The original material and/or application and its limitations
 - » New materials or processing used in practice (i.e., in manufacturing) and how and why it exceed original materials
 - » Limitations with the new materials/processing, if any
 - Conclusions and/or directions for future R&D for that material or application
 - Contribution of each member (e.g., literature survey, PowerPoint file preparation, others)
- **Dates:**
 - Date 1 (after 2nd exam): Group presentation topics due
 - Date 2 (last week of semester): oral presentation
- **Grouping:** each group should have 3 or 4 students, form group on your own
- **Grading:** 7 points maximum and all group members get the same grades
- **It needs to be at the COLLEGE LEVEL, quantitative, showing the concepts/knowledge learned in this class and not remaining at the middle-school or even primary school level!**



Contents Covered

- **Fundamental Concepts (chapter 1-5, 9, 10)**
 - Structure
 - Diffusion
 - Phase and phase transformation
- **Properties (Chapter 6-8, 17)**
 - Mechanical
 - Corrosion/degradation
- **Categories of materials (Chapter 10-16)**
 - Metals, Ceramics/Glass, Polymer, Composites: their processing and applications
- **Interrelationships between different aspects**



Types of Materials – By Chemistry/Composition

- **Metals:** Pure elements or compounds consist of metallic elements
 - Metallic bond
 - High thermal & electrical conductivity
 - Opaque, reflective
- **Ceramics:** – mostly solid compounds consists of at least one non-metallic elements (oxides, carbides, nitrides, sulfides)
 - Ionic or covalent bonding (refractory)
- **Polymers:** Large molecules contain carbon C as well as other elements such as H, O, N, Si, S, etc.
 - Usually covalent bond between atoms
 - Soft, ductile, low strength, low density, low melting point
- **Composites:** Consisting at least two distinctly different materials to realize optimized properties that combines the advantages of each components



Types of Materials – Other Classifications

By **application** or function

- Structural materials
- Electronic materials: Conductors; Semiconductors; Dielectrics
- Optical materials
- Magnetic materials
- Biomaterials
- ...

By certain **property**

- Electrical: conductor; insulator; semiconductor
- Hardness: hard versus “soft”
- Toughness: brittle versus ductile
- ...

By **source**

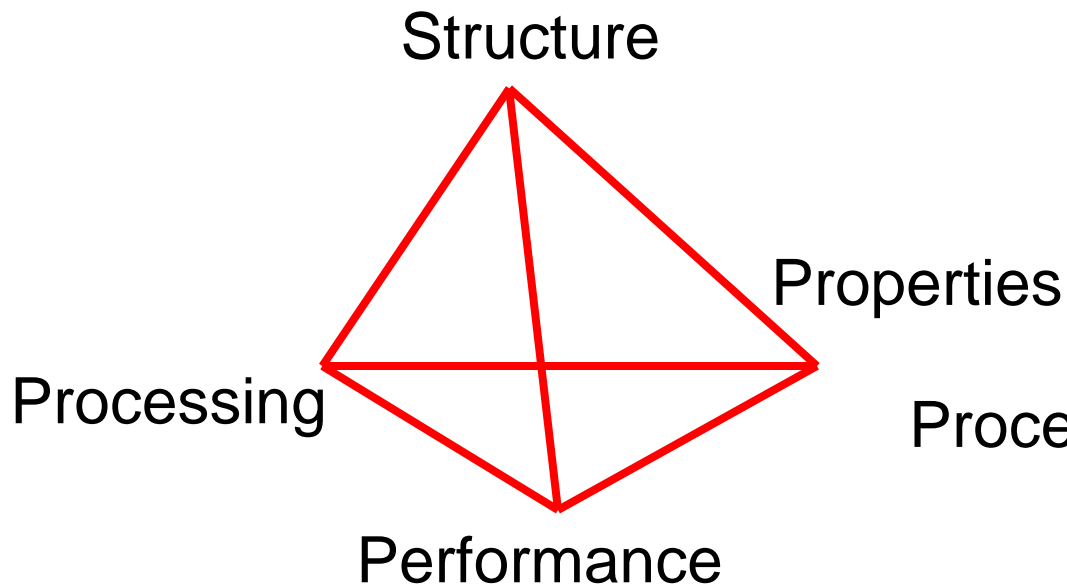
- Natural materials
- Man-made



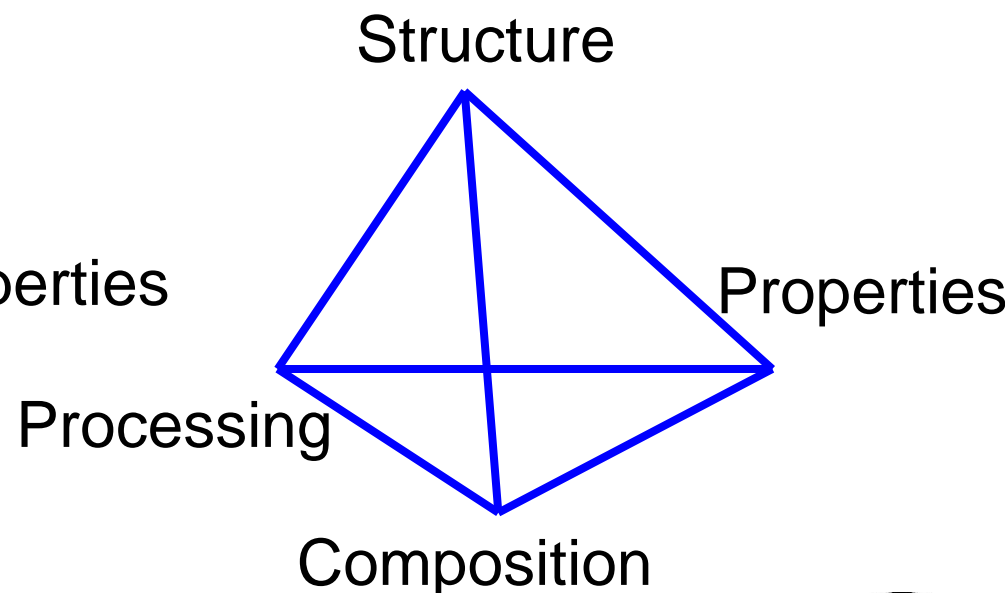
Materials Tetrahedron

- Basic “aspects” of materials engineering and their inter-relationships

Common “Materials” Tetrahedron



Dr. Cheng’s version of “Materials” Tetrahedron



Materials Composition

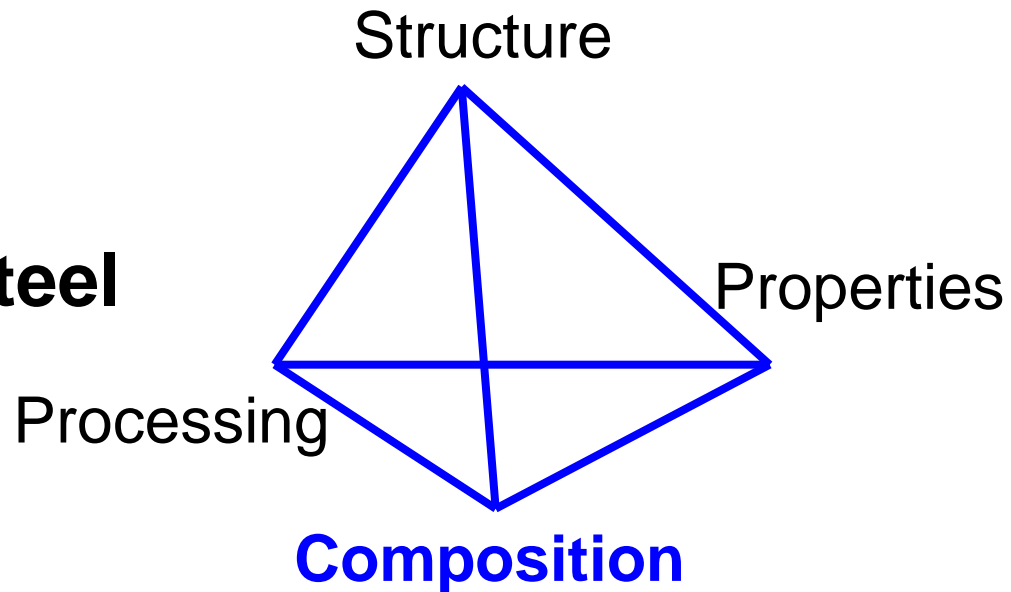
Chemistry

- Al vs. Ti vs. Fe
- Metal vs. Polymer
- ...

Concentration

- C content in iron & steel
- B or P content in Si
- ...

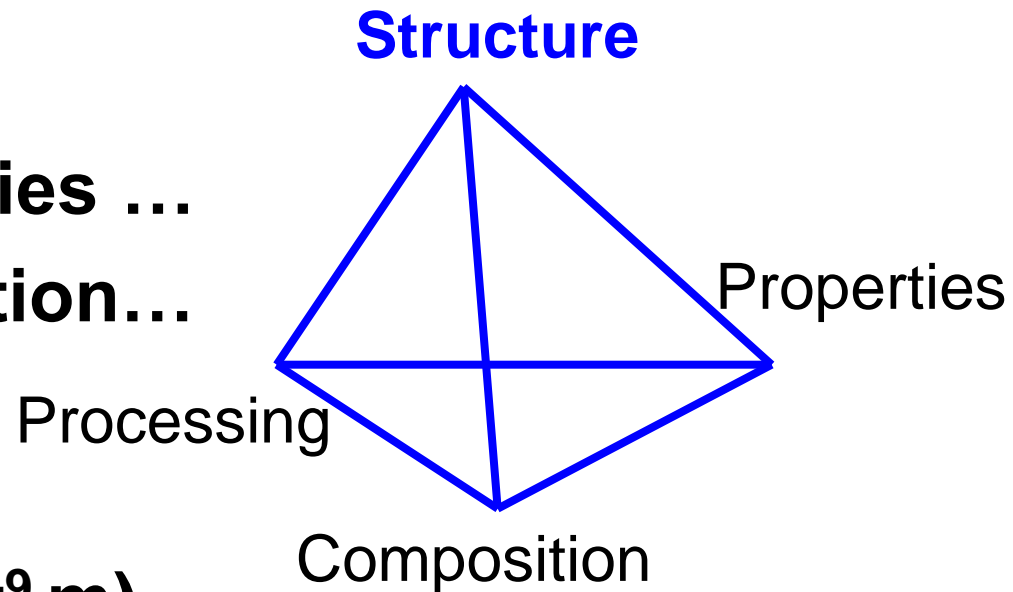
Dr. Cheng's version of "Materials" Tetrahedron



Materials Structure

- **Macro-scale** (e.g., millimeter, $\sim 10^{-3}$ m and above)
 - Bar, sheet, fiber...
- **Micro-scale** (e.g., micron, $\sim 10^{-6}$ m)
 - Grain and boundaries ...
 - Pores and distribution...
- **Atomic scale** (e.g., nanometer, $\sim 10^{-9}$ m)
 - Crystal (or lattice) structure

Dr. Cheng's version of "Materials" Tetrahedron



Materials Processing

Heat

- Casting
- Sintering
- Annealing
- ...

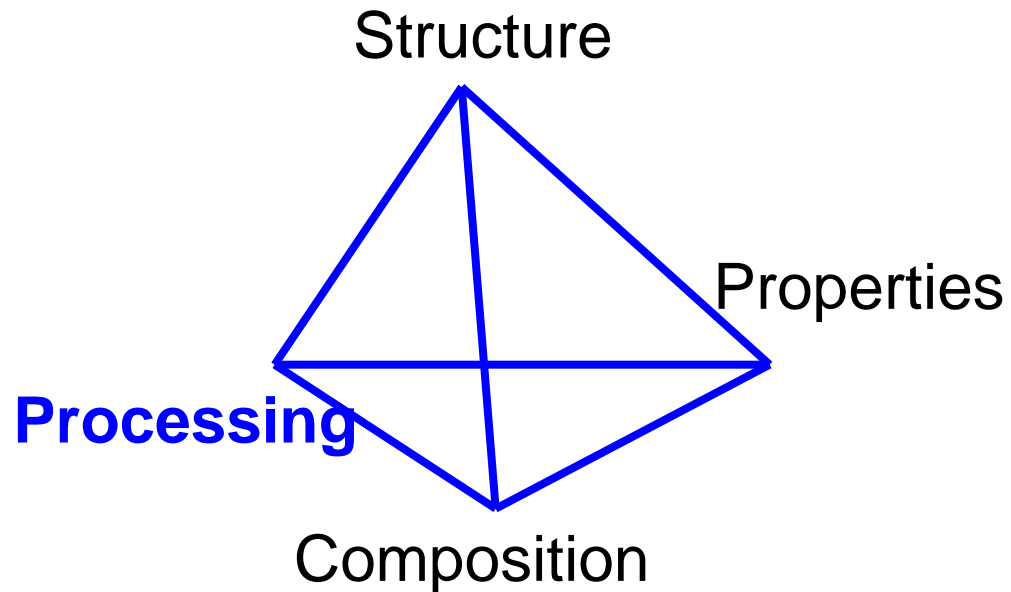
Mechanical

- Rolling
- Cutting and grinding
- ...

Other (Chemical, Electrical, etc.)

- Sputtering
- Electrodeposition
- ...

Dr. Cheng's version of "Materials" Tetrahedron



Materials Properties

Mechanical

Electrical

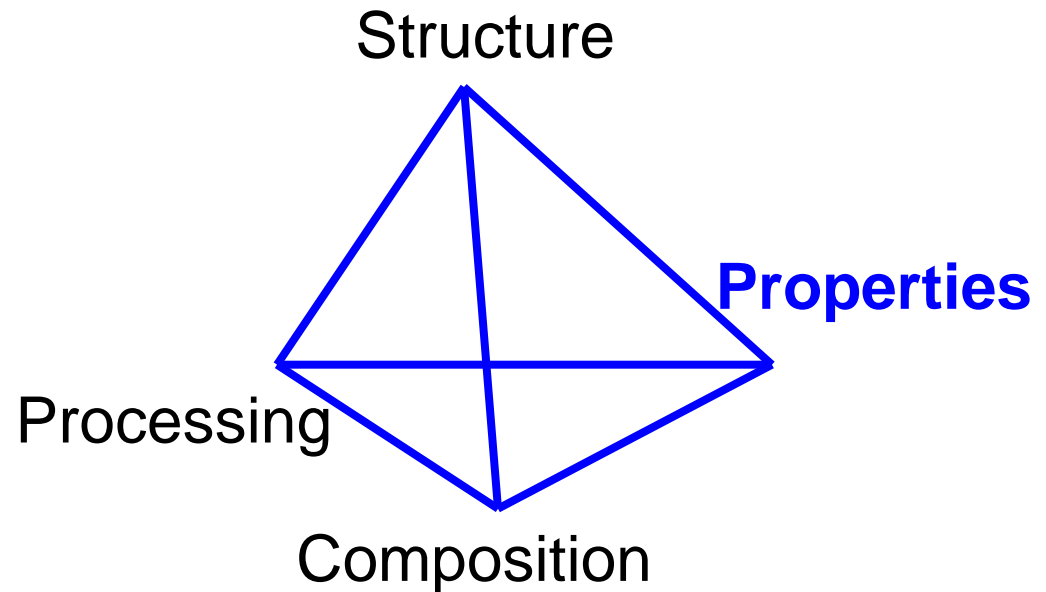
Thermal

Optical

Magnetic

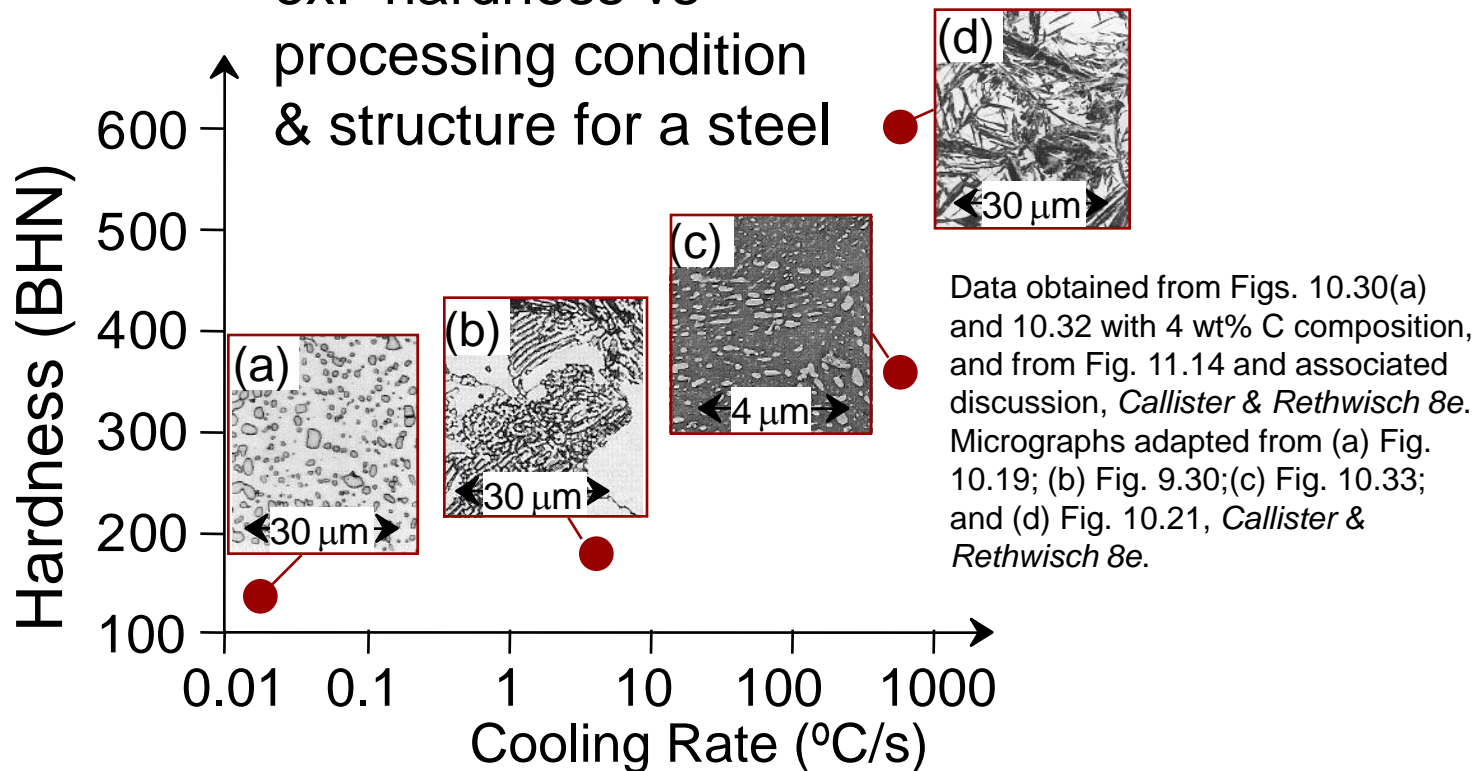
Other

Dr. Cheng's version
"Materials" Tetrahedron



Composition – Structure – Processing – Properties Relationships

- **Properties** largely depends on **composition**
ex: hardness of aluminum vs. diamond; steel vs. pure iron
- Once composition fixed, **properties** is also influenced by **(micro)structure**
ex: hardness vs



- **Processing** can change **structure** (and sometimes composition) and resulting properties



“Evolution” of Material Choice for a Particular Application

1. Pick an **application** → Determine required **Properties**
Properties: mechanical, electrical, thermal, magnetic, optical, deteriorative.
2. **Properties** → Identify/discover candidate **Material(s)**
Material: composition & structure
3. **Material** → Identify required **Processing**
Processing: changes *structure* and overall *shape*
ex: casting, sintering, vapor deposition, doping forming, joining, annealing.

Material of choice for a specific application may change/evolve overtime



“Evolution” of a Particular Material

1. Pick a material → Determine intrinsic **Properties**
Copper
2. **Applications** → Identify possible (new) applications
Mechanical tool; Weapon; Housewares and sculptures;
Structural elements such as tubings; Lead wires for currents;
Interconnect in semiconductor device
3. **Processing** → Identify required **Processing**
Processing: changes *structure* and overall *shape*
ex: Metallurgy, electrodeposition, casting, joining, annealing,
polishing,

Application & processing evolve overtime for a given material



Class Exercise

Materials selection

- Picks a specific engineering application and discuss the potential materials that might be used for that particular application:
 - *Example: lamp post: wood, concrete, steel, aluminum...*

Or

Materials evolution

- Pick a specific material and discuss its current and potential future applications
 - **Example: gold: coin, decoration/ jewelry, reactor, electrical wiring...**



SUMMARY

Course Goals:

- Understand the basic ways of classification of engineering materials
- Be aware of the different aspects (i.e, **composition**, **properties**, **structure**, and **processing**.) in material selection/design/manufacturing considerations
- Understand the concept of “evolution” for materials



Homework #1

- Read Chapters 1 and then give a statement in your homework confirming you finished the required reading
- >100 word short write-up describing a specific material or an application (for which material is the limiting factor) that interests you and explain why

